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Sertifikaat

REPUBLIEK VAN SUID AFRIKA

PATENT KANTOOR
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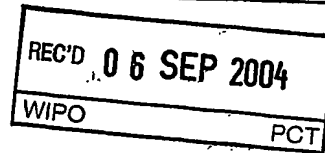


Certificate

REPUBLIC OF SOUTH AFRICA

PATENT OFFICE
DEPARTMENT OF TRADE AND
INDUSTRY

Hiermee word gesertifiseer dat
This is to certify that



the documents attached hereto are true copies of Forms P1, P2
and provisional specification and drawings of South African Patent Application
No. 2003/6374 in the name of Element Six (Proprietary) Limited

Filed : 15 August 2003

Entitled : Boron Coated

Abrasive Particles

Geteken te
Signed at **PRETORIA**

in die Republiek van Suid-Afrika, hierdie
in the Republic of South Africa, this

10th dag van
August 2004
day of

Registrar of Patents

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REPUBLIC OF SOUTH AFRICA		REGISTER OF PATENTS		PATENTS ACT, 1978		
OFFICIAL APPLICATION		LODGING DATE: PROVISIONAL		ACCEPTANCE DATE		
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INTERNATIONAL CLASSIFICATION		LODGING DATE: COMPLETE		GRANTED DATE		
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FULL NAME(S) OF APPLICANT(S)/PATENTEE(S)						
71	ELEMENT SIX (PROPRIETARY) LIMITED					
APPLICANTS SUBSTITUTED:				DATE REGISTERED		
71						
ASSIGNEE(S)				DATE REGISTERED		
71						
FULL NAME(S) OF INVENTOR(S)						
72	1. RAS, ANINE HESTER 2. DAVIES, GEOFFREY JOHN					
PRIORITY CLAIMED		COUNTRY	NUMBER	DATE		
N.B. Use International abbreviation for country (see Schedule 4)		33	NIL	31	NIL	
				32	NIL	
TITLE OF INVENTION						
54	BORON COATED ABRASIVE PARTICLES					
ADDRESS OF APPLICANT(S)/PATENTEE(S)						
DEBID ROAD, NUFFIELD, SPRINGS, 1559, GAUTENG, SOUTH AFRICA						
ADDRESS FOR SERVICE				S & F REF		
74	SPOOR & FISHER, SANDTON			PA135773/P		
PATENT OF ADDITION NO.			DATE OF ANY CHANGE			
61						
FRESH APPLICATION BASED ON			DATE OF ANY CHANGE			

APPLICATION FOR A PATENTAND ACKNOWLEDGEMENT OF RECEIPT
(Section 30 (1) - Regulation 22)

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HASR 711

The granting of a patent is hereby requested by the undermentioned applicant on the basis of the present application filed in duplicate

OFFICIAL APPLICATION NO.

REPUBLIC VAN SUID AFRIKA
S & F REFERENCE

PA135773/P

21 01 2003/6374

FULL NAME(S) OF APPLICANT(S)

71 ELEMENT SIX (PROPRIETARY) LIMITED

ADDRESS(ES) OF APPLICANT(S)

DEBID ROAD, NUFFIELD, SPRINGS, 1559, GAUTENG, SOUTH AFRICA

TITLE OF INVENTION

54 BORON COATED ABRASIVE PARTICLES

THE APPLICANT CLAIMS PRIORITY AS SET OUT ON THE ACCOMPANYING FORM P.2. THE EARLIEST PRIORITY CLAIM IS:

COUNTRY: NIL

NUMBER: NIL

DATE: NIL

THIS APPLICATION IS FOR A PATENT OF ADDITION TO PATENT APPLICATION NO.

21 01

THIS APPLICATION IS A FRESH APPLICATION IN TERMS OF SECTION 37 AND IS BASED ON APPLICATION NO.

21 01

THIS APPLICATION IS ACCOMPANIED BY:

- ☒ 1. A single copy of a provisional specification of 8 pages.
- ☐ 2. Drawings of sheets
- ☐ 3. Publication particulars and abstract (Form P.8 in duplicate).
- ☐ 4. A copy of Figure of the drawings (if any) for the abstract.
- ☐ 5. Assignment of invention.
- ☐ 6. Certified priority document.
- ☐ 7. Translation of the priority document.
- ☐ 8. Assignment of priority rights.
- ☐ 9. A copy of the Form P.2 and the specification of S.A. Patent Application No.
- ☐ 10. Declaration and power of attorney on Form P.3.
- ☐ 11. Request for ante-dating on Form P.4.
- ☐ 12. Request for classification on Form P.9.
- ☒ 13. Form P.2 in duplicate.
- ☐ 14. Other.

74 ADDRESS FOR SERVICE: SPOOR & FISHER, SANDTON

Dated: 15 August 2003


 SPOOR & FISHER
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REGISTRAR OF PATENTS DESIGNS,
TRADE MARKS AND COPYRIGHTRECEIVED
2003 AUG 15REGISTRATEUR VAN PATENTE, MODELLE,
HANDELSMERKE EN OUTEURSREG

REGISTRAR OF PATENTS

REPUBLIC OF SOUTH AFRICA
PATENTS ACT, 1978**PROVISIONAL SPECIFICATION**

(Section 30(1) – Regulation 27)

OFFICIAL APPLICATION NO.

21	01	2003/6378
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LODGING DATE

22	15 AUGUST 2003
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FULL NAMES OF APPLICANTS

71	ELEMENT SIX (PROPRIETARY) LIMITED
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FULL NAMES OF INVENTORS

72	RAS, ANINE HESTER DAVIES, GEOFFREY JOHN
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TITLE OF INVENTION

54	BORON COATED ABRASIVE PARTICLES
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BACKGROUND OF THE INVENTION

This invention relates to boron coated abrasives, a process for their production, and to boron coated abrasives for use in saw blade segments, grinding wheels and other abrasive-containing tools.

Abrasive particles such as diamond and cubic boron nitride are commonly used in cutting, grinding, drilling, sawing and polishing applications. In such applications, abrasive particles are mixed with metal powder mixes, then sintered at high temperatures to form bonded cutting elements. Typical bond matrices contain iron, cobalt, copper, nickel and/or alloys thereof.

Common problems in applications are retention of particles in the bond matrix, and resistance against oxidative attack during the sintering process and the subsequent application.

These problems are commonly addressed by coating the abrasive particles with metals or alloys which bond chemically to the particle, and alloy to the bond matrix. Typically, chemical vapour deposition (CVD) or physical vapour deposition (PVD sputter coating) techniques are used. A disadvantage of the CVD technique is that it often requires the use of highly toxic chemical compounds. A disadvantage of the PVD technique, being a low temperature technique, is that the metal coating does not form a chemical bond with the grit particles.

US 6,524,357 discloses a process for coating a superabrasive with a metal by heating the superabrasive and a coat-forming powder in an inert atmosphere. It is a requirement of this process that the coating powder contains a metal compound of which the metal is thermally reducible by the superabrasive.

The addition of boron oxide and/or boric acid to diamond powder to manufacture sintered diamond compacts at high pressure and high temperature in order to improve the strength of the compact, for use in cutting and drilling applications, is disclosed in JP 9142 932. Similarly, EP 0 352 811 describes the use of metal-coated superabrasives for the high pressure high temperature manufacture of superabrasive cutting elements and compacts.

US 4,902,652 discloses a method for producing diamond compacts at high pressure high temperature (i.e. in the region where diamond is stable) by using coated diamond particles. Although boron is mentioned as one of the coatings, only conventional techniques of ion sputtering and ion plating are disclosed.

SUMMARY OF THE INVENTION

A process for producing a boron coated abrasive includes the steps of contacting the abrasive to be coated with a boron source, at a coating temperature of about 800°C to about 1200°C in an inert atmosphere for a time sufficient to coat at least a portion of the abrasive.

The abrasive may be in any appropriate form including abrasive particles such as grit or granules, for example, larger abrasive bodies such as sheets or films of abrasive, for example, or even abrasive tools themselves such as cutting, grinding or polishing tools, for example.

In a preferred embodiment of the invention, abrasive particles are coated with boron by mixing the abrasive particles with a boron source, and heating the mixture at the coating temperature in an inert atmosphere for a time sufficient to coat at least a portion of the abrasive particles.

The boron source preferably comprises boron powder, in particular in combination with boric acid. The ratio of boron powder to boric acid is such as to optimise the available boron for coating the abrasive particles, and is typically 1:0 to 1:1, preferably about 1:0.7, by weight.

The final boron coat may be boron, boron carbide, boron oxide or a combination of any of these.

The process is preferably carried out at a temperature of about 800°C to about 1150°C, in particular at a temperature of about 1100°C to about 1150°C.

The process preferably includes a preheating step, which comprises heating the abrasive and boron source, which in the case of abrasive particles would typically be a mixture of abrasive particles and boron source, incrementally to a temperature of about 250°C to about 500°C, preferably about 300°C, and maintaining them at that temperature for a

period of about 15 minutes to about 45 minutes, in particular about 30 minutes, in order to soften and melt the boron source.

The abrasive and boron source are preferably incrementally heated to the coating temperature, typically at about 5°C/minute to about 15°C/minute, in particular about 10°C/minute, to a coating temperature of at least 800°C, preferably 1150°C.

The abrasive and boron source are preferably heated at the coating temperature for at least 30 minutes, in particular for 3 hours, most particularly for at least 6 hours.

The ratio of abrasive to boron source, particular with regard to abrasive particles, is preferably about 1:0.2 to about 1:20, in particular about 1:2, by weight.

The abrasive is preferably diamond or cubic boron nitride.

The invention extends to boron coated abrasives, in particular boron coated abrasive particles, preferably as manufactured by the process of the invention.

The invention also extends to a tool, in particular a cutting, grinding or polishing tool, which is either boron coated or comprises sintered boron coated abrasive particles in an appropriate bond matrix.

DESCRIPTION OF EMBODIMENTS

The present invention is directed at forming boron coated abrasives, in particular boron coated abrasive particles, which may be used in saw blade segments, grinding wheels and other grit-containing tools.

In a preferred embodiment of the invention, the abrasive is provided in the form of abrasive particles. However, it is to be understood that the coating method of the invention may be applied to any abrasives for use in grinding, cutting or polishing tools, or to such tools themselves. In the case of abrasive particles, these are mixed with boron and boric acid powder, in a suitable ratio, and furnaceed at a temperature between 800 and about 1200°C for a length of time in an inert atmosphere sufficient to coat the particles. After cooling, the coated abrasive particles are separated from the powders by any convenient method, such as screening, flotation or elutriation. The coated particles are washed to remove any adhering powder particles.

By an "inert" atmosphere is meant that the atmosphere is free of contaminants that may interfere with the process. An inert atmosphere may be achieved by passing an inert gas through the reaction vessel or by subjecting the reaction vessel to a vacuum.

Boron (a non-metal) offers significant advantages as a coating for abrasives. The boron substitutes for carbon in the diamond lattice, thereby increasing the resistance of the diamond to oxidative and chemical attack during saw segment manufacture. Boron also alloys with Co and diffuses into iron bonds to improve mechanical properties of the bond material, such as creep strength, for example.

Although the method described here is simple: grit is mixed with boron and boric acid powder, and heated to temperatures in the range of 800 to 1200°C for a period of time in an inert atmosphere, it is very effective. No toxic chemicals are used. There is also a surprising improvement observed in the coating obtained when the coating temperature is adjusted to 1150°C, which appears to be the optimum temperature for carrying out the process of the invention.

Where the abrasive is in a form which is too large to form a mixture with the boron source, any other appropriate method of contacting the abrasive with

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AVOIDANCE

the boron source may be applied. For instance, in the case of an abrasive tool, this may be immersed in the boron source to provide contact. Alternatively, the boron source may be provided in a separate holder or container and once the appropriate coating temperature is reached, the boron source is vaporized and is able to contact the abrasive in an appropriate manner to coat it.

The invention will now be illustrated by way of the following non-limiting examples.

Example 1

32g of boric acid (H_3BO_3) was added to 46g boron powder, and mixed in a Turbula mixer for 1.5 hours. 2g of this mix was added to 1g of diamond grit (ex Element Six, SDB1100, 35/45 US mesh) in a closed plastic container and shaken by hand for 3 minutes to mix. The mix was placed in an alumina boat in a tube furnace. The temperature was increased to 300°C in 1 hour (5°C/minute), held at 300°C for 0.5 hours, then increased to 1100°C in 1 hour 20 minutes (10°C/minute), held at 1100°C for 3 hours, then allowed to cool naturally. The powder was separated from the boron-coated diamond particles by screening and washing with water. Some particles were completely coated with matte grey coating, and some (111) faces were not coated at all.

Example 2

The same procedure was followed as in Example 1, except that the dwell time at 1100°C was doubled to 6 hours. The coating appeared slightly improved in terms of coverage, but there were still some uncoated (111) faces, and some entirely uncoated particles.

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The same procedure was followed as in Example 2, except that the dwell temperature was increased by 50°C to 1150°C. The resulting coating had a very rough texture, contained boron, boron oxide and boron carbide, and consisted of crystalline, star-shaped structures. The particles were completely covered, with excellent coverage even at the edges and corners of the particles. The thickness of the coating was in the range of 1-2µm.

DATED THIS 15th DAY OF AUGUST 2003

SPOOR & FISHER
APPLICANT'S PATENT ATTORNEYS

THE UNIVERSITY OF CHICAGO